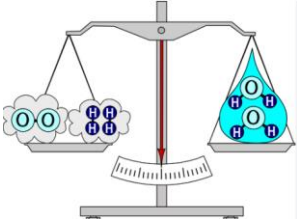


<p>catalyst</p> <p>cat / a / lyst</p>	<p>Substances that:</p> <p>Speed up chemical reactions.</p> <p>but</p> <p>Are unchanged at the end.</p>
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<p>combustion</p> <p>com / bus / tion</p>	<p>Chemical reaction in which:</p> <p>A substance reacts quickly with oxygen and gives out light and heat</p> <p>Also known as burning</p>
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<p>conservation of mass</p> <p>con / ver / sa / tion of mass</p>	<p>Chemical reaction where:</p>  <p>Total mass of reactants is equal to total mass of products.</p>
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<p>conserved</p> <p>con / served</p>	<p>When the quantity of something</p> <p>Does not change after a process takes place.</p>
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thermal decomposition

ther / mal de / com / po / si / tion

A chemical reaction in which

A compound breaks down on heating to form more than one product.